

# Lecture 2

Apr. 15, '24

As we discussed previously, we want to solve (each of the) radial Schrödinger equations

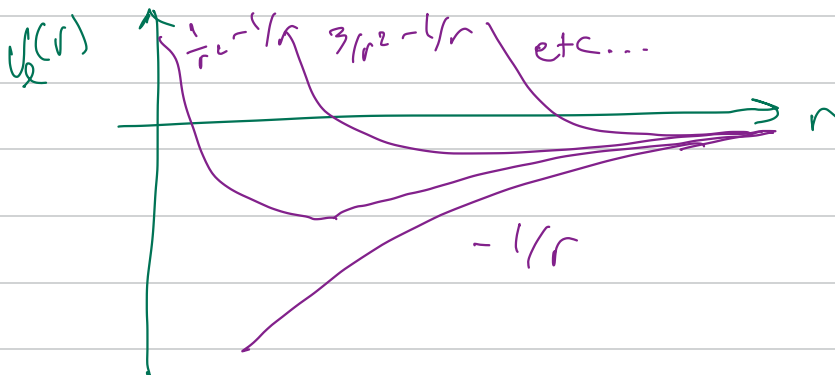
$$\left( -\frac{1}{2} \frac{d^2}{dr^2} + \overbrace{\frac{l(l+1)}{2r^2}}^{U_l(r)} - \frac{1}{r} - E_{nl} \right) u_{nl}(r) = 0$$

to obtain the energy levels <sup>H<sub>2</sub></sup> of hydrogen. This involves solving the (in principle infinitely many) 1D problems subject to boundary conditions ~

$$u_{nl} \sim r^{l+1}, \quad r \rightarrow 0$$

$$u_{nl} \sim 0, \quad r \rightarrow \infty \sim$$

with potential energies



We were gonna do this using a slightly exotic method: SuSy Q-Mech.

Let us begin with the generic 1D Ham:

$$H^{(1)} = -\frac{1}{2} \frac{d^2}{dx^2} + V^{(1)}(x) \quad \textcircled{1}$$

where  $V^{(1)}(x)$  has been shifted so that the ground state of  $H^{(1)}$  sits at zero energy, i.e.  $V^{(1)}(x) = V(x) - \epsilon_0$ .

This is a nice trick, since  $\psi_0(x)$ , the ground state, is nodeless  $\rightarrow$  so we have no pesky poles when trying to evaluate the following:

$$\begin{aligned} H^{(1)} \psi_0 &= 0 \\ -\frac{\psi_0''}{2} + V^{(1)} \psi_0 &= 0 \\ \rightarrow \underline{V^{(1)}} &= \frac{1}{2} \frac{\psi_0''}{\psi_0} \end{aligned}$$

That's cool!!

What other weird ways can we write  $V^{(1)}(x)$ ?  
How about -

$$V^{(1)}(x) = W(x)^2 - \frac{1}{\sqrt{2}} W'(x),$$

where  $W(x)$  is the SUPER potential!

... But why?

$I_+$ 's simple - we can use  $\tilde{W}$  to define  $\tilde{A}$ :

$$\tilde{A} = \frac{1}{\sqrt{2}} \frac{d}{dx} + W(x)$$

$$\tilde{A}^\dagger = -\frac{1}{\sqrt{2}} \frac{d}{dx} + W(x).$$

And these have the groovy property that:

$$H^{(1)} = A^\dagger A !$$

How? Just check:

$$\begin{aligned} H^{(1)}\psi &= \left(-\frac{1}{\sqrt{2}} \frac{d}{dx} + W\right) \left(\frac{1}{\sqrt{2}} \frac{d}{dx} + W\right) \psi \\ &= -\frac{1}{2} \frac{d^2}{dx^2} \psi - \frac{1}{\sqrt{2}} \left(W' \psi + W \psi'\right) + \frac{1}{\sqrt{2}} W \psi' + W^2 \psi \\ &= \left(-\frac{1}{2} \frac{d^2}{dx^2} + \underbrace{W^2 - \frac{1}{\sqrt{2}} W'}\right) \psi \end{aligned}$$

this was how we defined  $V^{(1)}$ !

$A$  has allowed us to factor any Ham!

But there's more! check out:

$$H^{(2)} = A A^\dagger.$$

$H^{(2)}$  has an eigenspectrum:

$$\begin{aligned} H^{(2)} \phi_n &= \epsilon_n \phi_n \\ \rightarrow A A^\dagger \phi_n &= \epsilon_n \phi_n \\ \rightarrow A^\dagger A (A^\dagger \phi_n) &= \epsilon_n (A^\dagger \phi_n) \\ \rightarrow A^\dagger A \tilde{\phi}_n &= \epsilon_n \tilde{\phi}_n \end{aligned}$$

Amazing!!

$H^{(1)}, H^{(2)}$  have  
degenerate  
spectra!!

Why does  $i$  matter? Notice that:

$$\begin{aligned}
 AA^\dagger \varphi &= \left( \frac{i}{\sqrt{2}} \frac{d}{dx} + W \right) \left( -\frac{i}{\sqrt{2}} \frac{d}{dx} + W \right) \varphi \\
 &= -\frac{1}{2} \varphi'' + \frac{i}{\sqrt{2}} (W' \varphi + W \varphi') - \frac{i}{\sqrt{2}} W \varphi' + W^2 \varphi \\
 &= -\frac{1}{2} \varphi'' + \underbrace{\left( \frac{W'}{\sqrt{2}} + W^2 \right)}_{V^{(1)}} \varphi = -\frac{1}{2} \varphi'' + \underbrace{V^{(2)}} \varphi
 \end{aligned}$$

And this is not the same as  $V^{(1)}$ !

$$\rightarrow H^{(1)} = -d^2/dx^2 + V^{(1)} \quad \text{and} \quad H^{(2)} = -d^2/dx^2 + V^{(2)}$$

Little aside:

$$\text{Since } V^{(1)}(x) = \frac{1}{2} \frac{\varphi_0''}{\varphi_0},$$

$$\text{and } V^{(2)}(x) = -\frac{W'}{\sqrt{2}} + W^2(x),$$

$$\text{we have } \frac{\varphi_0''}{\varphi_0} = -\sqrt{2} W'(x) + 2W(x).$$

$$\begin{aligned}
 \text{cool trick: } \left( \frac{\varphi_0''}{\varphi_0} \right) &= \left( \frac{\varphi_0'}{\varphi_0} \right)^2 + \left( \frac{\varphi_0'}{\varphi_0} \right)' \\
 &= \frac{\varphi_0'^2}{\varphi_0^2} + \varphi_0''/\varphi_0 - \varphi_0'^2/\varphi_0^2
 \end{aligned}$$

So:  $\frac{\varphi_0'}{\varphi_0}$  is:

$$\left( \frac{\varphi_0'}{\varphi_0} \right)^2 + \left( \frac{\varphi_0'}{\varphi_0} \right)' = \left( -\sqrt{2} W(x) \right)' + \left( -\sqrt{2} W(x) \right)^2$$

$$\text{So: } W(x) = -\frac{1}{\sqrt{2}} \frac{\varphi_0'}{\varphi_0} = -\frac{1}{\sqrt{2}} \frac{d}{dx} \ln \varphi_0(x).$$

We can solve for all  $\epsilon_n, \varphi_n$  of  $V^{(1)}$ , then use  $\varphi_0$  to get  $W$ , then  $W$  to get  $V^{(2)}$ , and its spectrum for free!!!!

to the degenerate spectrum!

The one exception is the zero-energy state.

Notice that if  $A^+A\psi_0=0$ ,  $\rightarrow A\psi_0=0$ .

This is a 1<sup>st</sup>-order DE,

$$\frac{1}{\sqrt{2}}\psi_0' + W(x)\psi_0 = 0, \text{ with an easy solution!}$$

$$\rightarrow \psi_0^{(1)} = \exp\left[-\int_{-\infty}^x W(y)dy\right]$$

This can be normalized

as long as  $\int_{-\infty}^x W(y)dy$  goes to  $\infty$  as  $x \rightarrow \infty$ .

So far, so good - so why can't  $AA^+\psi_0=0$  too?

This would require  $\psi_0(x) = \exp\left[+\int_{-\infty}^x W(y)dy\right]$   
↑ now a +, not -!!

These cannot be simultaneously normalized!

**PUNCHLINE:**

$H^{(1)} = A^+A$  and  $H^{(2)} = AA^+$  have

The same spectrum even though they correspond to very different potentials!

And:  $\psi_0$  is always obtained from a simple first order DE!

Let's do some examples. First, take

$$W(x) = ax^3.$$

This gives:

$$\bullet V^{(1)}(x) = a^2 x^6 - \frac{3}{\sqrt{2}} ax^2$$

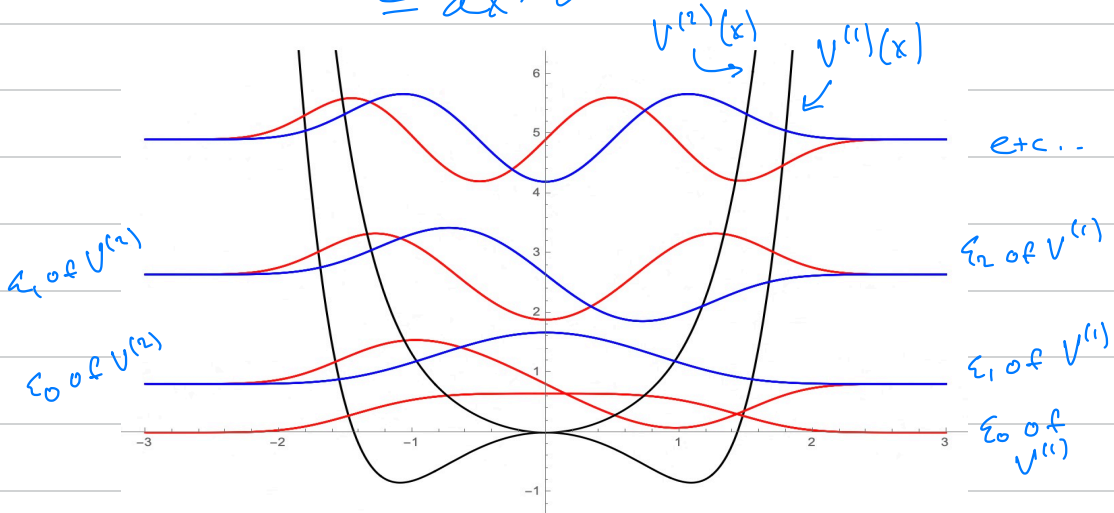
$$\bullet V^{(2)}(x) = a^2 x^6 - \frac{3}{\sqrt{2}} ax^2$$

$$\bullet \psi_0(x) = N \exp\left[-\frac{\sqrt{2}}{4} ay^4\right] \Big|_{-\infty}^x$$

$$= \underbrace{N e^{-\infty}}_{\text{normalization}} \exp\left(-\frac{\sqrt{2}}{4} ax^4\right)$$

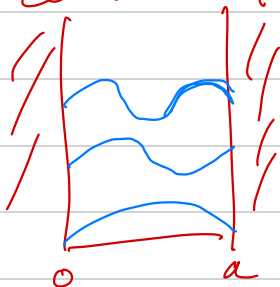
$$\bullet W = -\frac{1}{\sqrt{2}} \frac{d\psi_0'}{\psi_0} = -\frac{1}{\sqrt{2}} \left(-\frac{4}{4} ax^3 \sqrt{2}\right) e^{-} / e^{-}$$

$$= ax^3 \checkmark$$



This shows that two very different pots can have the same spectrum! But a little abstract...

For something a bit more down to earth, consider the infinite square well.



this has solutions  
 $\sim \sin(kx)$  where

$$k = n\pi/a \rightarrow E = \frac{k^2}{2} = \frac{n^2\pi^2}{2a^2}.$$

The SUPERPOT is thus:

$$W(x) = -\frac{1}{2} \frac{\pi}{a} \cot\left(\frac{\pi x}{a}\right).$$

(set  $a = \pi$  to  
 make the  
 rest easy..)

And this gives 2 potentials...

$$V^{(1,2)}(x) = W^2 \pm \frac{1}{2} W'(x)$$

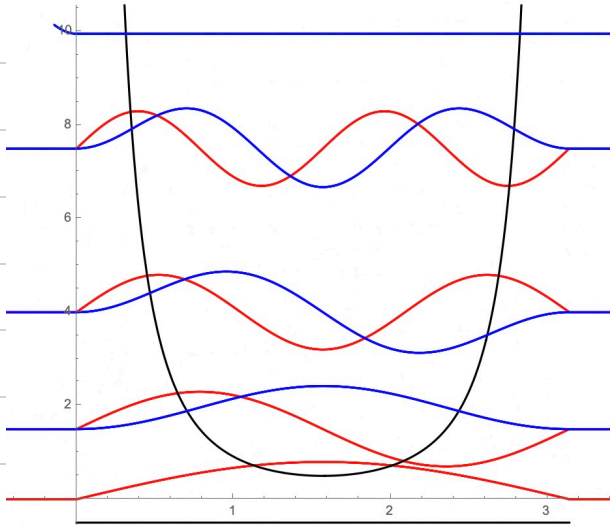
$$= \frac{1}{2} \cot^2(x) \mp \frac{1}{2} (-\csc^2(x))$$

$$= \frac{1}{2} \left( \frac{\cos^2(x) \mp 1}{\sin^2(x)} \right)$$

$$= -\frac{1}{2} \text{ or } \frac{1}{2} (2 \csc^2(x) - 1).$$

→ the first term gives the inf. sq. well, shifted by  $E_0$ . Cool! And the second is something else - something weird - but something

w/ spectrum  $\frac{n^2\pi^2}{2a^2}$ ,  $n = 2, 3, \dots$



Here we see  
the states of  
both systems...

This is a  $\longrightarrow$   
good review article  
if you want to  
learn more!

Physics Reports 251 (1995) 267-385

PHYSI

## Supersymmetry and quantum mechanics

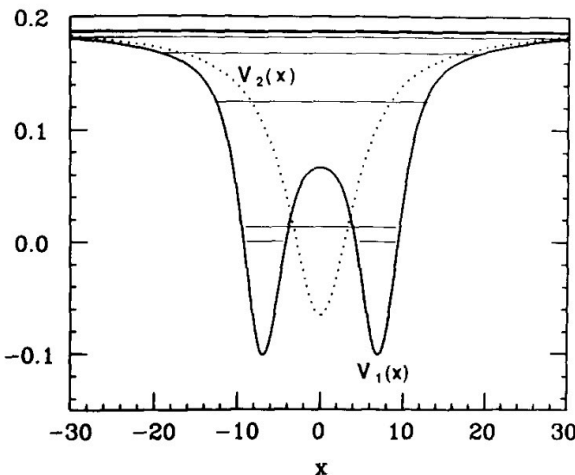
Fred Cooper<sup>a</sup>, Avinash Khare<sup>b</sup>, Uday Sukhatme<sup>c</sup>

<sup>a</sup> Theoretical Division, Los Alamos National Laboratory, Los Alamos, NM 87545, USA

<sup>b</sup> Institute of Physics, Bhubaneswar 751005, India

<sup>c</sup> Department of Physics, University of Illinois at Chicago, Chicago, IL 60607, USA

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e.g. how to solve  
wacky susy systems  
like this!



But now it's time to get back to our goal of solving the hydrogen atom:

$$H\psi_e = E_e \psi_e,$$

$$\text{where } H = -\frac{1}{2} \frac{d^2}{dr^2} + \frac{l(l+1)}{2r^2} - \frac{1}{r}.$$

First off, let's figure out our SUPPORT!

$$\text{From } V^{(l)} = W^2 - \frac{1}{\sqrt{2}} W' = \frac{l(l+1)}{2r^2} - \frac{1}{r} - E_0$$

we can guess the solution:

$$W(r) = C - D/r$$

$$\rightarrow \frac{l(l+1)}{2r^2} - \frac{1}{r} - E_0 = \frac{C^2}{r^2} - \frac{2CD}{r} + \frac{D^2}{r^2} - \frac{1}{\sqrt{2}} \frac{D}{r^2}$$

$$\text{So: } \underbrace{E_0 = -C^2}_A, \quad \underbrace{-2CD = -1}_B, \quad \underbrace{D^2 - D/\sqrt{2} = \frac{l(l+1)}{2}}_C$$

$$\text{From } \underline{C}: \quad D = \frac{l+1}{\sqrt{2}}$$

$$\rightarrow \text{From } \underline{B}: \quad C = \frac{1}{\sqrt{2}(l+1)}$$

$$\rightarrow \text{From } \underline{A}: \quad E_0 = -\frac{1}{2(l+1)^2}$$

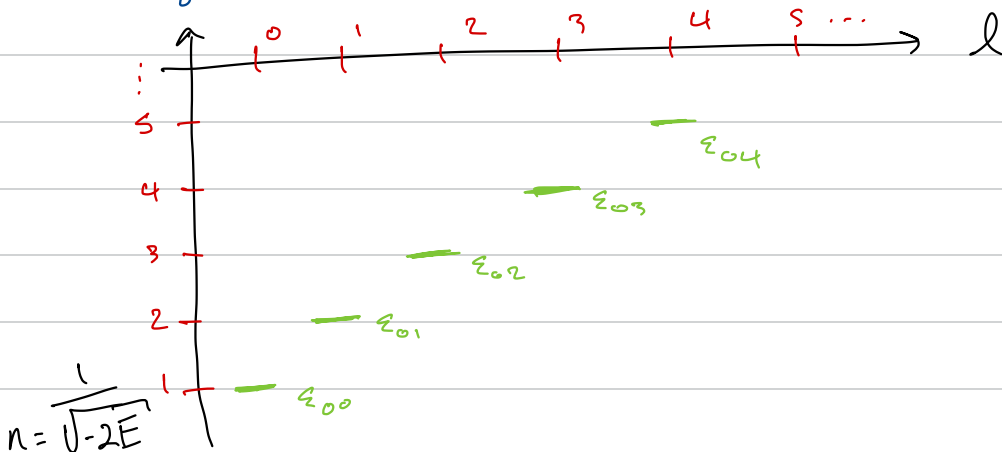
Wow! We already found the ground state energy for each  $l$ !!!

As a sanity check, let's compute the wf:

$$\begin{aligned}
 \psi_{0l}(r) &= e^{-\int_0^r \sqrt{2} W(y) dy} \\
 &= \exp\left[-\int_0^r \frac{1}{2+l} - \frac{(2+l)}{r} dr\right] \\
 &= \exp\left[-r/(2+l) + (2+l) \ln r\right] \\
 &= \frac{r^{2+l} e^{-r/(2+l)}}{\phantom{r^{2+l} e^{-r/(2+l)}}}, \text{ up to normalization.}
 \end{aligned}$$

Notice that the expected short- and long-range behavior holds, and this is a normalizable function! Hopefully, one we recognize from Kindergarten-QM!

Here's a nice way to visualize the gs energies we've obtained:



We've also determined

$$W(r) = \frac{1}{\sqrt{2}} \left[ \frac{l}{l+1} - \frac{l+1}{r} \right]$$

With  $W$ , we can compute the  $A_\ell$ 's:

$$\hat{A}_\ell = \frac{1}{\sqrt{2}} \left( \frac{d}{dr} + \frac{l}{2+1} - \frac{l+1}{r} \right)$$

$$\hat{A}_\ell^\dagger = \frac{1}{\sqrt{2}} \left( -\frac{d}{dr} + \frac{l}{2+1} - \frac{l+1}{r} \right)$$

These give us the Hams:

$$\begin{aligned} \bullet H_\ell^{(1)} &= A_\ell^\dagger A_\ell = \frac{1}{2} \left( -\frac{d^2}{dr^2} + \frac{\boxed{(l+1)l}}{r^2} - \frac{2}{r} + \frac{1}{(l+1)^2} \right) \\ &\quad \text{this is the!} \\ &= H_e^{\text{Coulomb}} - \underbrace{\left( -\frac{l}{2(l+1)^2} \right)}_{\substack{\text{E}_{0\ell} \\ \text{l-dependence} \\ \text{is normal}}} \quad \textcircled{2} \end{aligned}$$

$$\begin{aligned} \bullet H_\ell^{(2)} &= A_\ell A_\ell^\dagger = \frac{1}{2} \left( -\frac{d^2}{dr^2} + \frac{\boxed{(l+1)(l+2)}}{r^2} - \frac{2}{r} + \frac{1}{(l+1)^2} \right) \\ &= H_{\ell+1}^{\text{Coulomb}} + \frac{1}{2(l+1)^2} \quad \textcircled{1} \end{aligned}$$

Sanity check #2:  $H_\ell^{(1)}$  is the correct Ham, consisting of the Coulomb radial Ham less the ground state energy.

Notice that (1) leads us to:

$$H_{\ell+1}^{(1)} = H_{\ell+1}^{\text{Coulomb}} - \left( -\frac{1}{2(\ell+2)^2} \right)$$
$$\rightarrow H_{\ell+1}^{\text{Coul}} = H_{\ell+1}^{(1)} - \frac{1}{2(\ell+2)^2} \quad (2)$$

From (1) and (2) together, we have:

$$\underline{H_{\ell}^{(2)} = H_{\ell+1}^{(1)} + \frac{1}{2(\ell+1)^2} - \frac{1}{2(\ell+2)^2}} \quad (3)$$

Now we want to use these to generate the (as yet unknown) excited states.

• Let (3) act on the gs. of  $H_{\ell+1}^{(1)}$ :

$$H_{\ell}^{(2)} \phi_{0,\ell+1}^{(1)} = \left( \underbrace{H_{\ell+1}^{(1)}}_{=0} + \frac{1}{2(\ell+1)^2} - \frac{1}{2(\ell+2)^2} \right) \underbrace{\phi_{0,\ell+1}^{(1)}}_{\text{is the gs.}}$$

$$\rightarrow H_{\ell}^{(2)} \phi_{0,\ell+1}^{(1)} = \left( 0 + \frac{1}{2(\ell+1)^2} - \frac{1}{2(\ell+2)^2} \right) \phi_{0,\ell+1}^{(1)}$$

So:  $\star$  is an eval of  $H_{\ell}^{(2)}$ .

So far, so good!!

But by SUSY, non-zero evals of  $H_e^{(2)}$  are shared by  $H_e^{(1)}$ !

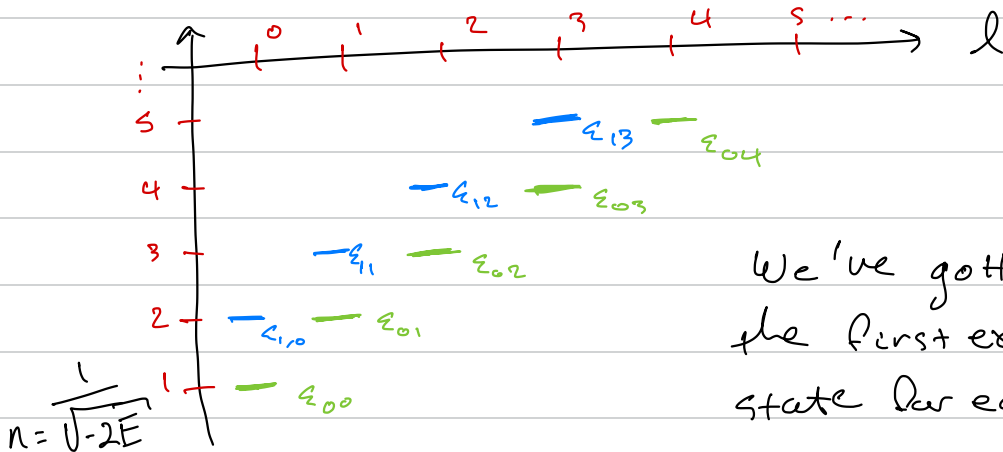
$$\rightarrow A_e^+ H_e^{(2)} \phi_{l, l+1}^{(1)} = \star A_e^+ \phi_{l, l+1}^{(1)}$$

$$\rightarrow \underbrace{A_e^+ A_e}_{H_e^{(1)}} \underbrace{(A_e^+ \phi_{l, l+1}^{(1)})}_{\phi_{l, l}^{(1)}} = \star \underbrace{(A_e^+ \phi_{l, l+1}^{(1)})}_{\phi_{l, l}^{(1)}}$$

$$\rightarrow H_e^{(1)} \phi_{l, l}^{(1)} = \star \phi_{l, l}^{(1)}$$

So:  $\star = \frac{1}{2(l+1)^2} - \frac{1}{2(l+2)^2}$  is the corresponding eigenvalue of state  $\phi_{l, l}^{(1)}$ !

And thus, from (1):  $\epsilon_{l, l}^{\text{Coulomb}} = -\frac{1}{2(l+2)^2}$



We've gotten the first excited state for each  $l$ !

Let's go around the loop one more time!

We act on the first excited state of  $H_{l+1}^{(1)}$  with  $H_l^{(2)}$ :

$$H_l^{(2)} \phi_{1,l+1}^{(1)} = \left( \underbrace{H_{l+1}^{(1)} + \frac{1}{2(l+1)^2} - \frac{1}{2(l+2)^2}}_{\frac{1}{2(l+2)^2} - \frac{1}{2(l+3)^2}, \text{ see page above}} \right) \phi_{1,l+1}^{(1)}$$

$$= \frac{1}{2} \left( \frac{1}{(l+2)^2} - \frac{1}{(l+3)^2} + \frac{1}{(l+1)^2} - \frac{1}{(l+2)^2} \right) \phi_{1,l+1}^{(1)}$$

So:  $H_l^{(2)} \phi_{1,l+1}^{(1)} = A_l A_l^\dagger \phi_{1,l+1}^{(1)} = \left[ \frac{1}{2(l+1)^2} - \frac{1}{2(l+3)^2} \right] \phi_{1,l+1}^{(1)}$

→ multiply by  $A_l^\dagger \dots$

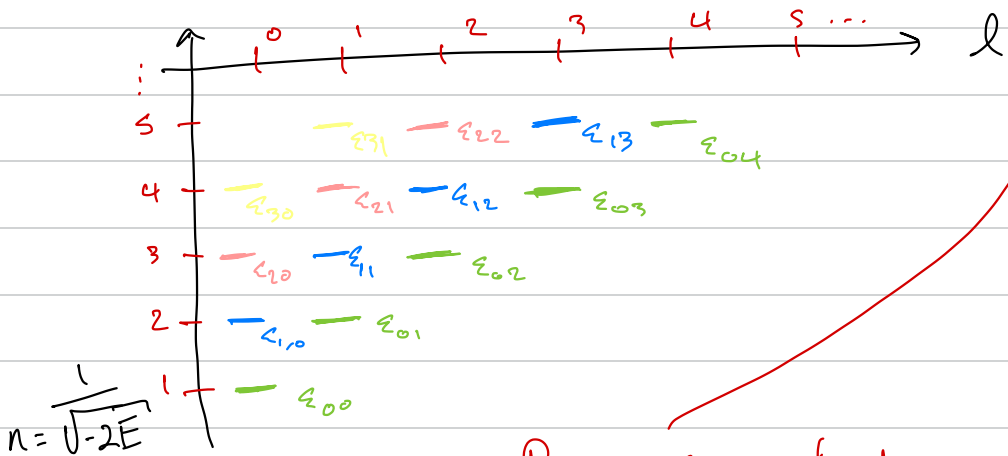
$$\rightarrow H_l^{(1)} \underbrace{[A_l^\dagger \phi_{1,l+1}^{(1)}]}_{\phi_{2,l}^{(1)}} = \left[ \frac{1}{2(l+1)^2} - \frac{1}{2(l+3)^2} \right] \phi_{2,l}^{(1)}$$

Subtracting the offset energy gives:

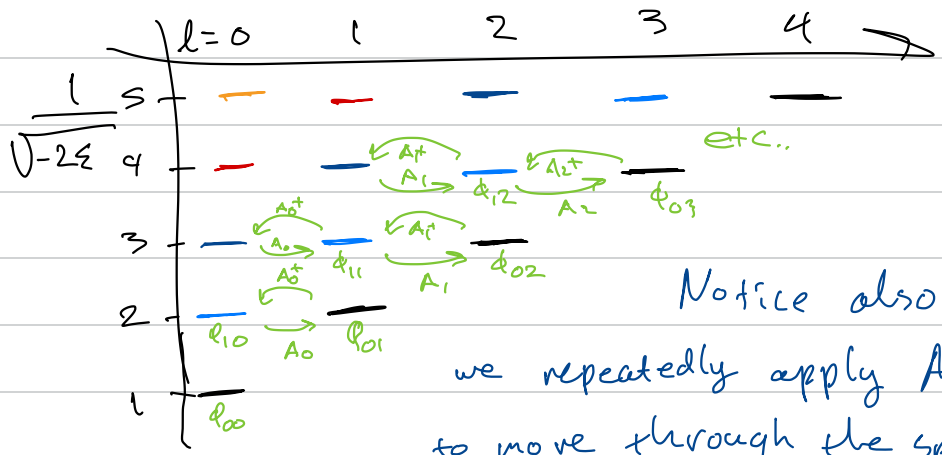
$$H_l^{\text{Coulomb}} \phi_{2,l}^{(1)} = -\frac{1}{2(l+3)^2} \phi_{2,l}^{(1)}$$

Looks like we have a pattern!

$$\epsilon_{n,l} = -\frac{1}{2(n+l+1)^2}$$



Recognize that we can relabel  $n+l \rightarrow n$ , an integer, and hence recover the same Rydberg formula as before.



Notice also how we repeatedly apply  $A$  or  $A^+$  to move through the spectrum!

This procedure we just did is part of a much larger context of 'shape invariant potentials', a category of 1D potentials defined by the relationship

$$V^{(2)}(x, a_1) = V^{(1)}(x, a_2) + R(a_1)$$

↓ indep. of x

$$a_2 = f(a_1)$$

This lets us define a family of Hamiltonians

$$H_s = -\frac{1}{2} \frac{d^2}{dx^2} + V_1(x, a_s) + \sum_{k=1}^{s-1} R(a_k)$$

$$a_s = f(f(f \dots (a_1))) \dots$$

Who cares? Well,

$$H_{s+1} = -\frac{1}{2} \frac{d^2}{dx^2} + \underbrace{V_1(x, a_{s+1}) + \sum_{k=1}^s R(a_k)}_{= V_2(x, a_s) + R(a_s)}$$

$$\rightarrow = -\frac{1}{2} \frac{d^2}{dx^2} + V_2(x, a_s) + \sum_{k=1}^{s-1} R(a_k)$$

$H_s$  and  $H_{s+1}$  are SUSY HAMILTONS  $\rightarrow$  so their spectra are identical except for  $H_s$ 's GS, w/  $\epsilon_0^s = \sum_{k=1}^{s-1} R(a_k)$ .



Doing this for all  $s$  values gives the spectrum

$$E_n = \sum_{k=1}^n R(a_k), \quad E_0 = 0.$$

Check all this w/ Mr. Coulomb!

—

In conclusion, we used SUSY HAMS to solve the hydrogen problem. This introduced a wider concept of factorizable Hamiltonians and SIPs, which are another way of categorizing all analytically solvable systems!

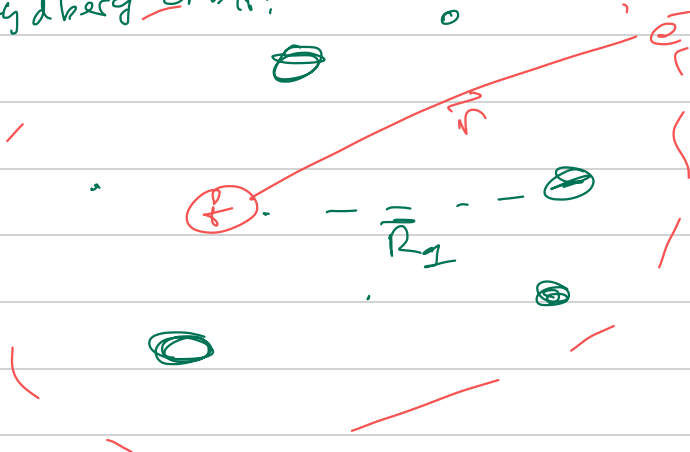
Historically, Mr. Schrö himself was the first to factorize the Coulomb problem!

For more reading:

Cooper, Khare, & Sukhatme *Phys. Rep.* 251, 267 (1995)

Before closing the book on SUSY HAMS, here's one more connection to Rydberg physics.

Consider a Rydberg atom in a dense gas, where many ground state atoms lie inside the Rydberg orbit:



The interaction of the Ryd  $e^-$  w/ each gs atom can be very well approx'd by

$$U(\vec{r}, \vec{R}_q) = 2\pi a \delta^3(\vec{r} - \vec{R}_q)$$

where  $a =$  the zero energy scattering length.

This was first studied by Fermi in 1934 (we will say more about this later).

To make a long story short: we want to see how these gs atoms perturb the Ryd atom with principal QN  $n$ .

The  $l$  states are degenerate, so we must diagonalize  $\hat{V} = 2\pi a \sum_{\mathbf{g}=1}^N S(\vec{r} - \vec{R}_{\mathbf{g}})$

$$\begin{aligned} \rightarrow V_{n\ell m, n\ell m'} &= 2\pi a \sum_{\mathbf{g}=1}^N \int \phi_{n\ell m}^*(\vec{r}) S^3(\vec{r} - \vec{R}_{\mathbf{g}}) \phi_{n\ell m'}(\vec{r}) \\ &= 2\pi a \sum_{\mathbf{g}=1}^N \phi_{n\ell m}^*(R_{\mathbf{g}}) \phi_{n\ell m'}(R_{\mathbf{g}}). \end{aligned}$$

This is an  $n^2 \times n^2$  matrix - yikes to diagonalize!

But wait! Define:  $A^\dagger = \phi_{n\ell m}^*(R_{\mathbf{g}})$   
 $A = \phi_{n\ell m}(R_{\mathbf{g}})$

These are 2 rectangular matrices...

$$A^\dagger = \begin{matrix} & 1, 2, \dots, N \\ \begin{matrix} \vdots \\ \vdots \\ \vdots \end{matrix} & \boxed{\phantom{\text{matrix}}} \\ n^2 & \end{matrix}$$

$$A = \begin{matrix} & 1, 2, \dots, n^2 \\ \begin{matrix} \vdots \\ \vdots \\ \vdots \end{matrix} & \boxed{\phantom{\text{matrix}}} \\ N & \end{matrix}$$

All that SUSY stuff says:

$$\underline{V} = 2\pi a A^\dagger A$$

$$\rightarrow \underline{V} \vec{c} = \epsilon \vec{c} = A^\dagger A \vec{c}$$

$$\rightarrow \epsilon(A \vec{c}) = \underline{AA^\dagger} (A \vec{c})$$

we only need this  $N \times N$  matrix!

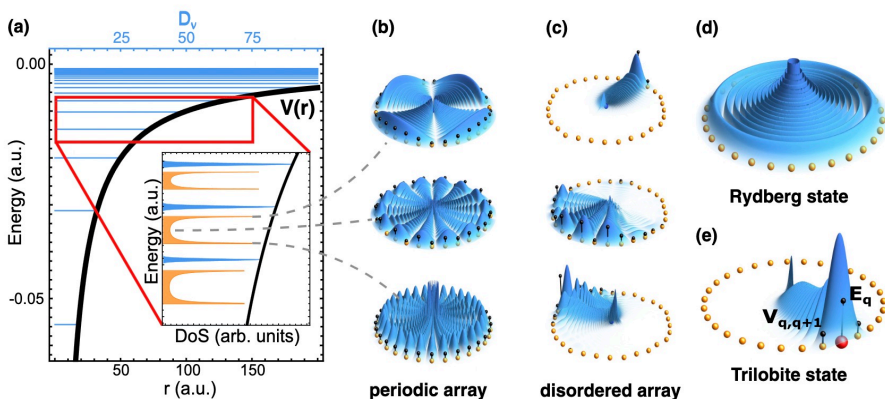
Not only is this numerically way easier, it is conceptually very cool as it links two SUSY-like Hamiltonians!

$$H_1 = -\frac{\hbar^2}{2m} \nabla^2 + \frac{1(l+1)}{2r^2} - \frac{1}{r} + 2\pi a \sum_{\mathbf{q}=1}^N \delta^3(\vec{r}-\vec{R}_{\mathbf{q}})$$

(a perturbed Rydberg atom)

$$H_2 = \sum_{\mathbf{q}=1}^N E_{\mathbf{q}} |\mathbf{q}\rangle \langle \mathbf{q}| + \sum_{\mathbf{q}, \mathbf{q}'}^N V_{\mathbf{q}\mathbf{q}'} |\mathbf{q}\rangle \langle \mathbf{q}'|$$

(a tight-binding lattice!)



See MTE, Eisfeld, and Rost Phys. Rev. Research **5**, 033032 (2023)

Anderson localization of a Rydberg  $e^-$ !

One more "application":

U. Kostelecký + M.M. Nieto,

"Evidence for a Phenomenological  
Supersymmetry in Atomic Physics"

PRL 53 2285 (1984).

The authors use the  $V^{(1)}$  and  $V^{(2)}$  we derived for the Coulomb Hamiltonian, w/  $V^{(2)}$  having the same spectrum as  $V^{(1)}$ , minus the  $gS$ .

But then they say that this is like an atom w/o the  $1s$  orbitals, i.e.  $Li$ :

$Li: 1s^2 \underline{2s}$

the spectrum of these states  
should resemble those of  $H$ !

And, the same thing goes for  $Na$

$Na: 1s^2 2s^2 3s \dots$

They claim in the end that similarities between the excited state energies of different atomic species imply a kind of evidence for supersymmetry!

Can this possibly be true? Stay tuned!

In conclusion, in this lecture we learned:

- How to factorize a Hamiltonian

$$H = A^\dagger A$$

so that its SuSy-partner

$$\tilde{H} = A A^\dagger$$

could be found

- That  $H$  and  $\tilde{H}$  have degenerate spectra except for the ground state.
- How to use this to:
  - relate one  $H$  w/ another to solve 2 problems for the price of 1.
  - Build up the solution of the Coulomb problem directly.

Although this SuSy stuff won't be too useful in general applications, we can actually use it now to introduce one major topic of this course,

Quantum Defect Theory!

This example comes from:

U. Kostelecky + M.M. Nieto,  
"Evidence for a Phenomenological  
Supersymmetry in Atomic Physics"  
PRL S3 2285 (1984)

These authors start from the SUSY partner  
potentials that we used,

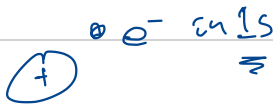
$$V^{(1)} = -\frac{1}{r} + \frac{l(l+1)}{2r^2} + \frac{1}{2(l+1)^2}$$

$$V^{(2)} = -\frac{1}{r} + \frac{(l+1)(l+2)}{2r^2} + \frac{1}{2(l+1)^2},$$

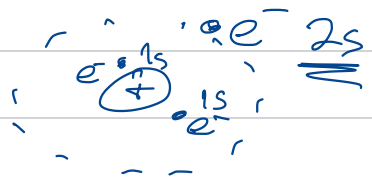
and then claim: since  $H^{(2)} = -\frac{1}{2} \frac{d^2}{dr^2} + V^{(2)}$  has  
the same spectrum as  $H^{(1)} = -\frac{1}{2} \frac{d^2}{dr^2} + V^{(1)}$  but  
with the ground state removed, then  $H^{(2)}$   
describes a system with the  $1s$  orbital removed.

So: in the absence of electron-electron  
interactions,  $H^{(2)}$  is the Hamiltonian of Li!

H:



Li:





The authors then go on to claim that this means that atomic spectra contain evidence for supersymmetry! (And that this supersymmetry is broken by e-e interactions.)

How do they justify this claim? With experiment, of course!

TABLE I. Energy differences (in units of  $10^4 \text{ cm}^{-1}$ ) between selected levels of the alkali-metal atoms (Ref. 6). The best agreements with the H levels are underlined and are those that one would expect from the discussion in the text.  $\rightarrow$  those where e-e

n	H		Li		Na		d
	s/p/d	s	p	d	s	p	
8-9	0.0360	0.0416	0.0366	<u>0.0360</u>	0.0607	0.0496	<u>0.0362</u>
6-7	0.0808	0.0980	0.0826	<u>0.0810</u>	0.164	0.124	<u>0.0814</u>
4-5	0.245	0.329	0.255	<u>0.247</u>	0.746	0.477	<u>0.249</u>
3-4	0.533	0.781	0.555	<u>0.534</u>	2.57	1.33	<u>0.538</u>
2-3	1.52	2.72	<u>1.60</u>	...	...	...	...

not  
is  
weak.

Although agreement between the underlined transitions seems pretty good, you might find this line of argument a bit fishy... and you should!

To see why, we'll need to solve the Coulomb problem a few more times...